

## Brief information about the project

Name of the project	AP09058570 «The development of the electrochemical modification method of carbon sorbent to impart the required sorption properties»
Relevance	Currently there are many processes and commercialized technologies available for the removal of toxic substances from drinking water and industrial wastewater, such as chemical precipitation, adsorption, biosorption, ion exchange, reverse osmosis, membrane filtration, electrochemical treatment, coagulation and flocculation, solvent extraction, cathodic electrodeposition, cementation processes, etc. However, despite the ease of use, selectivity, low time consumption, etc., these methods have many disadvantages in the form of large capital investments, high energy and operating consumption, sludge release, as well as the selection of conditions and materials for selectivity process. The novelty of this project lies in endowing the surface of a carbon material with the required sorption properties by changing the acid-base properties of the surface after modification.
Purpose	Development of a method for electrochemical and chemical influence on carbon material to form a modified functional layer on its surface for carrying out a selective sorption process
Objectives	<ol style="list-style-type: none"><li>1) obtaining carbon materials based on plant and carbon-mineral raw materials;</li><li>2) carrying out modification of carbon materials under various conditions and studying its effect on surface properties;</li><li>3) manufacturing an electrolyzer to carry out the process in dynamics;</li><li>4) carrying out electrochemical modification while varying the electrolyte composition and modes for further study of the sorption of organic and inorganic substances.</li></ol>
Expected and achieved results	Carbon sorbents based on mineral and plant raw materials were obtained, and their physicochemical characteristics were studied. Electrochemical modifications of the resulting sorbents were carried out under different conditions in different electrolytes. It has been demonstrated that electrochemical oxidation of activated carbon sorbents leads to improved adsorption and morphological characteristics. An increase in the number of functional groups after anodic modification of activated carbon has a direct effect on the sorption properties. During the study, a new electrolyzer was developed and constructed for the modification of activated carbon sorbents.
Research team members with their identifiers (Scopus Author ID, Researcher ID, ORCID, if available) and links to relevant profiles	Atchabarova Azhar aidarovna <a href="https://www.scopus.com/authid/detail.uri?authorId=56998822600">https://www.scopus.com/authid/detail.uri?authorId=56998822600</a> <a href="https://www.webofscience.com/wos/author/record/53537705">https://www.webofscience.com/wos/author/record/53537705</a> <a href="https://orcid.org/0000-0002-4600-2728">https://orcid.org/0000-0002-4600-2728</a>  Abduakhytova Dinara Aktaykyzy <a href="https://www.scopus.com/authid/detail.uri?authorId=57344630000">https://www.scopus.com/authid/detail.uri?authorId=57344630000</a> <a href="https://www.webofscience.com/wos/author/record/GYA-5917-2022">https://www.webofscience.com/wos/author/record/GYA-5917-2022</a>

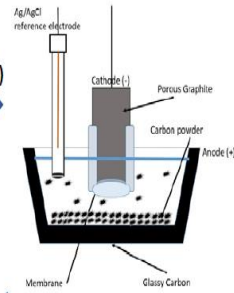
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List of publications with links to them	<p>1) Saken Abdimomyn, Azhar Atchabarova*, Dinara Abduakhytova, Rustam Tokpayev, Kanagat Kishibayev, Tamina Khavaza, Andrey Kurbatov, Graziella Liana Turdean, Mikhail Nauryzbayev. Investigation of the functional layer formation on the surface of carbon material // Studia UBB Chemia, 67 (4), 2022. DOI:10.24193/subbchem.2022.4.10</p> <p>2) Azhar Atchabarova, Saken Abdimomyn*, Dinara Abduakhytova*, Kanagat Kishibayev, Yelena Zlobina, Andrey Kurbatov, Graziella Liana Turdean, Thierry Djenizian. Electrochemical modification of the carbon materials surface by hydroxyl groups // Journal of Solid State Electrochemistry, 2023 <a href="https://doi.org/10.1007/s10008-023-05780-8">https://doi.org/10.1007/s10008-023-05780-8</a></p>
Patents	not planned



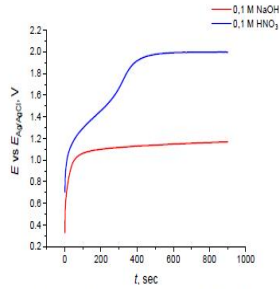
HTC (240°C, 24 hours)  
SGA (800°C, 1 hour)



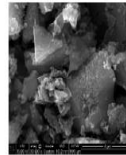
Electrochemical modification (EcM)  
 $j = 10 \text{ mA/cm}^2; t = 15 \text{ min}$



*Mechanism of electrochemical modification*

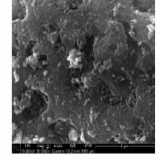


EcM  
0,1 M NaOH

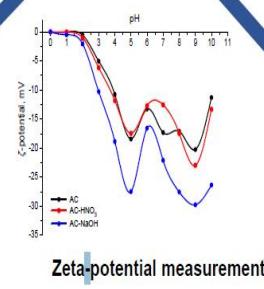


AC-NaOH

EcM  
0,1 M HNO<sub>3</sub>



AC-HNO<sub>3</sub>



Zeta-potential measurements

*Mechanism of copper ion adsorption on modified activated carbon*

